# NASA TECHNICAL NOTE



NASA TN D-3135



# HALL COEFFICIENT OF TANTALUM CARBIDE AS FUNCTION OF CARBON CONTENT

by Gilbert J. Santoro and R. T. Dolloff
Lewis Research Center
Cleveland, Ohio

NATIONAL AERONAUTICS AND SPACE ADMINISTRATION - WASHINGTON, DECEMBER 1965



# HALL COEFFICIENT OF TANTALUM CARBIDE AS FUNCTION OF CARBON CONTENT

By Gilbert J. Santoro and R. T. Dolloff

Lewis Research Center Cleveland, Ohio

NATIONAL AERONAUTICS AND SPACE ADMINISTRATION

For sale by the Clearinghouse for Federal Scientific and Technical Information Springfield, Virginia 22151 — Price \$1.00

### HALL COEFFICIENT OF TANTALUM CARBIDE AS FUNCTION OF CARBON CONTENT

# by Gilbert J. Santoro and R. T. Dolloff

#### Lewis Research Center

#### SUMMARY

The Hall coefficient within the entire cubic phase region of tantalum carbide was investigated by a direct-current method and was found to be negative and to increase in absolute magnitude with increasing carbon content, while the electrical resistivity was found to decrease. These results, as well as the known variation of the magnetic susceptibility and the superconductivity transition temperature with carbon content, indicate that a one-carrier model is not adequate to describe the entire system. A composition in the ditantalum carbide phase region was also found to have a negative Hall coefficient, its magnitude being greater than that of any of the other samples. Magnetoresistivities were too small to be accurately measured.

#### INTRODUCTION

A number of investigators (refs. 1 to 4) have measured the Hall coefficients of the group IV and V transition metal carbides and nitrides and their solid solutions having sodium chloride type crystal structure in order to understand more fully the electronic binding in these materials. Their samples have been restricted either to nearly stoichiometric compositions or to solid solutions of nearly stoichiometric components. The results of these investigations have been expressed in terms of the variation of the valence electron concentration (VEC), that is, the valence electrons per formula with the expressed or implied assumption that the electronic band structure is not strongly affected by differences in the ionic cores. A recent band structure calculation on titanium carbide (ref. 5), however, implies that substitutional alloying of either the metal or the nonmetal atom may significantly alter the density-of-states curve, even if the VEC is unchanged. The histograms of density of states for titanium carbide and titanium nitride as recently calculated by Ern and Switendick (ref. 6) clearly differ from each other not only with respect to the position of the Fermi level, but also with respect to the energy distribution of the density-of-states and to the degree of band interaction. Thus, the band

structure of one of these compounds cannot be universally employed for the others. These calculations, then, confirm the desirability of varying the VEC without resorting to substitutional alloying. Since these compounds exhibit rather wide single-phase regions, the VEC can be varied over a large range merely by changing the concentration of the nonmetal atom. Thus, in order to study more fully the effect of changes in the VEC on the electrical nature of tantalum carbide (TaC), room temperature Hall measurements were made on TaC ribbons of different carbon contents. The electrical resistivity of each specimen was also determined, and attempts were made to determine magnetoresistivity. Measurements were also made on a sample of the hexagonal carbide, Ta<sub>2</sub>C. The data are discussed in relation to the band structure determinations which have appeared in the literature for the group IV and group V transition metal carbides and nitrides and in conjunction with other properties of nonstoichiometric tantalum carbide.

## EXPERIMENTAL PROCEDURE

Tantalum carbide ribbons were prepared by heating pieces of tantalum foil (19 by 0.32 by 0.0025 cm) in measured amounts of propane for 6 hours at  $2000^{\circ}$  C. The purity of the tantalum was at least 99.89 percent. The method is similar to that used previously in the preparation of substoichiometric tantalum carbide filaments (ref. 7). An X-ray diffraction pattern was made on each of the specimens with a 114.6-millimeter-diameter Debye-Scherrer camera using copper  $K\alpha$  radiation. The lattice parameters,  $a_0$ , were calculated by means of a least-square extrapolation on an IBM 7094 computer. With these data and an established literature curve of  $a_0$  against carbon content (ref. 8) the

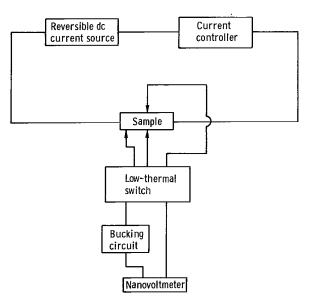


Figure 1. - Schematic of electrical circuit of Hall apparatus.

composition of each sample was ascertained. The estimated accuracy of the carbon determination using this procedure is about  ${\rm TaC}_{X\pm0.006}.$ 

A 1.4-centimeter length from each carburized ribbon was used as the Hall specimen. A direct-current method was employed in the Hall measurements. The electrical circuit is shown in figure 1. All the vital circuit parts were electrically and magnetically shielded. The low thermal switch permitted the selection of the desired pair of leads while introducing thermal voltages of less than 10<sup>-9</sup> volt. All stray voltages could be suppressed by the bucking

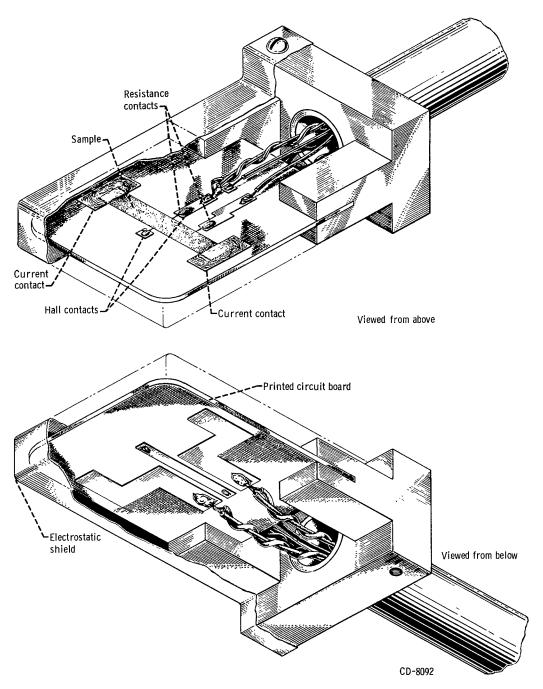


Figure 2. - Schematic of Hall probe.

circuit to less than  $10^{-9}$  volt. The voltage signal over a 10-minute interval was stable to within  $10^{-9}$  volt.

A drawing of the probe head, which was 4.4 by 2.5 by 0.6 centimeter, is shown in figure 2. The printed circuit board provided a convenient way of minimizing circuit loops, which could otherwise produce stray voltages and fields. The leads were connected to the copper strips on the circuit board with solder that has a low thermoelectric effect with copper. The electrical contacts to the sample were provided by a liquid mercury-indium alloy to circumvent problems of wetting and generation of thermal stresses when other methods of soldering or welding were attempted.

Measurements were taken of the transverse voltage with reversal of both current and magnetic field. First the voltage was measured as a function of current (fig. 3(a)) at the highest magnetic field available. The linear variation in these curves indicated freedom from thermal gradients caused by excessive Joule heating (proportional to the current squared). Then a convenient current was chosen, and the transverse voltage was measured as a function of the magnetic field (fig. 3(b)). This procedure indicated the validity of the linear variation of the Hall voltage with the magnetic field as given in the Hall equation

$$E_{H} = \frac{RBI}{t} \times 10^{-8}$$

where

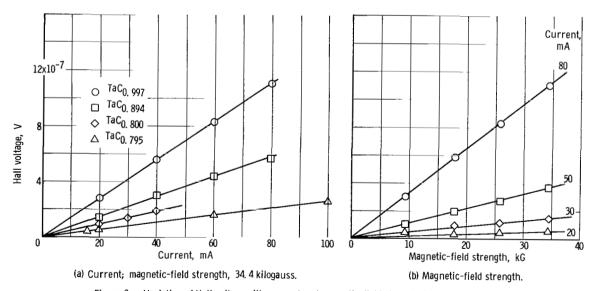


Figure 3. - Variation of Hall voltage with current and magnetic-field strength for various compositions.

E<sub>H</sub> Hall voltage, V

R Hall coefficient, cc/C

B magnetic-field strength, G

I current, A

t sample thickness, cm

The existence of isothermal conditions was further verified by remeasuring a specimen in high-thermal-conductivity oil (high-grade transformer oil) and obtaining the same Hall coefficient.

#### RESULTS

The Hall coefficient was found to be negative for all compositions tested, indicating electrons as the predominant carriers. Within the TaC phase region the Hall coefficient

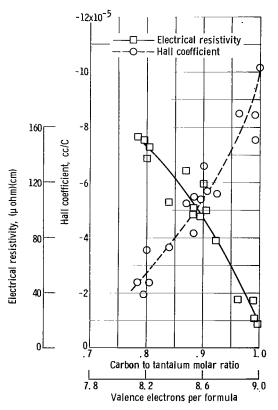


Figure 4. - Room-temperature variation of resistivity and Hall coefficient as functions of carbon content for cubic tantalum carbide,

became more negative with increasing carbon content (see fig. 4). For stoichiometric TaC a value of about  $-10\times10^{-5}$  cubic centimeter per coulomb was found, as compared with literature values of  $-11\times10^{-5}$  cubic centimeter per coulomb (refs. 1, 3, and 4) and  $-6.4\times10^{-5}$  cubic centimeter per coulomb (ref. 9). The latter value came from poorly characterized materials. The electrical resistivity data of the cubic carbide agreed with those previously reported (ref. 7), decreasing with increasing carbon content (see fig. 4). The magnetoresistivity was too small to be measured accurately, an upper limit of about  $10^{-6}$  being typical under an applied field of 34.4 kilogauss.

For the  ${\rm Ta_2C}$  specimen, the Hall coefficient was -15.5×10<sup>-4</sup> cubic centimeter per coulomb, an order of magnitude more negative than any value derived from the cubic samples. No values have been reported in the literature for this ditantalum carbide. A value of 102 microhm centimeters was found for the electrical resistivity of the  ${\rm Ta_2C}$  sample. Here again

the magnetoresistivity was too small to be measured accurately.

#### DISCUSSION

It has been assumed that the band structure of the IVth and Vth group transition metal carbides and nitrides are all similar (refs. 10 and 11), and it has been reported that the IVth group carbides exhibit Hall values of an order of magnitude more negative than those for the Vth group (refs. 1 to 4). On the basis of this knowledge and with the previous assumption applied to the present study, the Hall coefficient of TaC would be predicted to become more negative with decreasing carbon content and to approach the value for group IV carbides at a VEC of about 8.0, that is, at TaC<sub>0.75</sub>. The observed variation is opposite to this prediction; as shown in figure 4, the Hall coefficient becomes less negative as carbon content is decreased. The failure of this prediction casts doubt on the validity, even as a first approximation, of applying the band structure of any one of these materials to the entire group by a mere adjustment of the Fermi level in accordance with the VEC.

The findings reported here also show that the one-carrier model cannot be utilized for nonstoichiometric compositions of tantalum carbide, since, based on such a model, the Hall coefficient for the low-carbon samples yields carrier densities which are greater than normal metallic carrier densities. Thus, at  ${\rm TaC}_{0.75}$ , the number of conduction electrons would be about  $4\times10^{23}$  per cubic centimeter as compared, for example, to about  $0.8\times10^{23}$  per cubic centimeter for silver (ref. 12). Furthermore, this electron density for  ${\rm TaC}_{0.75}$  corresponds to about 9 electrons per formula, whereas its VEC is only 8.

The same conclusion can be reached from the following consideration: the variation of the magnetic susceptibility (refs. 13 and 14) and the superconductivity transition temperature (ref. 15) with carbon content in the cubic tantalum carbide phase indicates an increase in the density of states with increased carbon concentration. The electrical conductivity similarly increases. Thus, the number of conduction electrons increases, and this rise, on the basis of a one-carrier model, should be reflected by a decrease in the Hall coefficient, a prediction contrary to the variation in figure 4.

While the one-carrier scheme has been employed to explain the data collected from refractory transition metal carbides and nitrides of stoichiometric composition (refs. 1 and 3) and from their solid solutions (refs. 2 and 3), it is clear that such a scheme is not adequate for nonstoichiometric compositions of cubic tantalum carbide. However, before a unique electrical transport model can be assigned to this system, the factors involved in the conduction process must be ascertained. It is believed that the carbon vacancies greatly influence the conductivity in nonstoichiometric tantalum carbide. It

is important, therefore, to separate this contribution to the conductivity, by means of low-temperature measurements, and to determine its dependence upon composition.

### CONCLUSIONS

The room-temperature electrical transport properties of nonstoichiometric tantalum carbides are dominated by negative carriers throughout the entire cubic phase region. A one-carrier model is not adequate to explain the data, and low-temperature measurements are required before a specific model can be proposed.

Lewis Research Center,
National Aeronautics and Space Administration,
Cleveland, Ohio, September 29, 1965.

#### REFERENCES

- 1. L'vov, S.N.; Nemchenko, V.F.; and Samsonov, G.V.: Effect of Nonmetallic Atoms on the Electric Properties of High-Melting Compounds of Transition Metals. Soviet Powder Met. and Metal Ceramics, vol. 4, no. 10, July-Aug. 1962, pp. 231-236.
- Itoh, Fumitake; Tsuchida, Takashi; and Takaki, Hideo: Hall Effect of the Solid Solutions of TiC<sub>x</sub>N<sub>1-x</sub> and Ti<sub>x</sub>V<sub>1-x</sub>C. J. Phys. Soc. Japan, vol. 19, no. 1, Jan. 1964, pp. 136-137.
- 3. Piper, John: Electrical Properties of Some Transition-Metal Carbides and Nitrides. Rept. No. C-21, Union Carbide Res. Inst., Apr. 1964.
- 4. Samsonov, G.V.; und Paderno, V.N.: Über die Herstellung von Karbidmischkristaller und die Untersuchung ihrer Physikalischen Eigenschaften. (Concerning the Preparation of Carbide Solid Solutions and the Investigation of Their Physical Properties.)
  Planseeber. Pulvermet., bd. 12, heft 1, Apr. 1964, pp. 19-31.
- 5. Williams, Wendill S.; and Lye, Robert G.: Research to Determine the Mechanisms Controlling the Brittle-Ductile Behavior of Refractory Cubic Carbides. Rept. No. ML-TDR-64-25, Pt. II, Union Carbide Corp., Mar. 1965.
- 6. Ern, V.; and Switendick, A.C.: Electronic Band Structure of TiC, TiN, and TiO. Phys. Rev., vol. 137, no. 6A, Mar. 15, 1965, pp. 1927-1936.
- 7. Santoro, Gilbert: Variation of Some Properties of Tantalum Carbide with Carbon Content. AIME Trans., vol. 227, Dec. 1963, pp. 1361-1368.

- 8. Bowman, Allen L.: The Variation of Lattice Parameter with Carbon Content of Tantalum Carbide. J. Phys. Chem., vol. 65, no. 9, Sept. 27, 1961, pp. 1596-1598.
- 9. Tsuchida, Takashi; Nakamura, Yoji; Mekata, Mamoru; Sakurai, Junji; and Takaki, Hideo: Hall Effect in the Carbides of Transition Metals. J. Phys. Soc. Japan, vol. 16, no. 12, Dec. 1961, pp. 2453-2456.
- 10. Bilz, Heinz: Über Elektronenzustände von Hartstoffen mit Natriumchloridstruktur. (Concerning the Electronic Structure of Hard Materials with the Sodium Chloride Structure.) Z. Physik, bd. 153, heft 1, 1958, pp. 338-358.
- 11. Dempsey, E.: Bonding in the Refractory Hard-Metals. Phil. Mag., ser. 8, vol. 8, no. 86, Feb. 1963, pp. 285-299.
- 12. Mott, N.F.; and Jones, H.: The Theory of the Properties of Metals and Alloys. Dover Pub., Inc., 1958, p. 284.
- 13. Bittner, H.; and Goretzki, H.: Magnetic Investigation of the Carbides TiC, ZrC, HfC, VC, NbC and TaC. Monatsh. Chem., vol. 93, no. 5, 1962, pp. 1000-1004.
- 14. Santoro, G.: Magnetic Susceptibilities of Compositions in Tantalum-Tantalum Carbide System. NASA TN D-2638, 1965.
- 15. Giorgi, A.L.; Szklarz, E.G.; Storms, E.K.; Bowman, Allen L.; and Matthias, B.T.: Effect of Composition on the Superconducting Transition Temperature of Tantalum Carbide and Niobium Carbide. Phys. Rev., vol. 125, no. 3, Feb. 1, 1962, pp. 837-838.

2/12/85

"The aeronautical and space activities of the United States shall be conducted so as to contribute . . . to the expansion of human knowledge of phenomena in the atmosphere and space. The Administration shall provide for the widest practicable and appropriate dissemination of information concerning its activities and the results thereof."

-NATIONAL AERONAUTICS AND SPACE ACT OF 1958

## NASA SCIENTIFIC AND TECHNICAL PUBLICATIONS

TECHNICAL REPORTS: Scientific and technical information considered important, complete, and a lasting contribution to existing knowledge.

TECHNICAL NOTES: Information less broad in scope but nevertheless of importance as a contribution to existing knowledge.

TECHNICAL MEMORANDUMS: Information receiving limited distribution because of preliminary data, security classification, or other reasons.

CONTRACTOR REPORTS: Technical information generated in connection with a NASA contract or grant and released under NASA auspices.

TECHNICAL TRANSLATIONS: Information published in a foreign language considered to merit NASA distribution in English.

TECHNICAL REPRINTS: Information derived from NASA activities and initially published in the form of journal articles.

SPECIAL PUBLICATIONS: Information derived from or of value to NASA activities but not necessarily reporting the results of individual NASA-programmed scientific efforts. Publications include conference proceedings, monographs, data compilations, handbooks, sourcebooks, and special bibliographies.

Details on the availability of these publications may be obtained from:

SCIENTIFIC AND TECHNICAL INFORMATION DIVISION

NATIONAL AERONAUTICS AND SPACE ADMINISTRATION

Washington, D.C. 20546